# Studies of Force Induced Changes in Single Molecule Systems

- Prentiss Group
- Nelson Group at Harvard
- Whitesides Group

# Why do Single Molecule Experiments?

- Bulk measurements give only average values
  - Heterogeneities are masked
  - Intermediate states are difficult to detect

# What sorts of Single Molecule Force Experiments are there?

#### Adhesion Measurements

- Use force to pull apart single molecule bonds
  - » Ligand-Receptor Studies
  - » DNA unzipping

#### Elasticity Experiments

#### - Use force to stretch single molecules

» dsDNA and ssDNA stretching provides structure and free energy information

Receptor: A molecular structure or site on the surface or interior of a cell that binds with substances such as hormones, antigens, drugs, or neurotransmitters A <u>molecular structure within</u> a <u>cell</u> or on the <u>surface</u> characterised by selective <u>binding</u> of a <u>specific</u> <u>substance</u> and a specific <u>physiologic effect</u> that accompanies the binding, for example, cell surface receptors for <u>peptide hormones</u>, <u>neurotransmitters</u>, <u>antigens</u>, <u>complement</u> <u>fragments</u> and <u>immunoglobulins</u> and <u>cytoplasmic</u> receptors for <u>steroid hormones</u>. 2. A <u>sensory nerve terminal</u> that <u>responds</u> to <u>stimuli</u> of various kinds.

Ligand: An ion, a molecule, or a molecular group that binds to another chemical entity to form a larger complex Any <u>molecule</u> that <u>binds</u> to another, in <u>normal usage</u> a <u>soluble</u> molecule such as a <u>hormone</u> or <u>neurotransmitter</u>, that binds to a <u>receptor</u>. The <u>decision</u> as to which is the ligand and which the receptor is often a <u>little arbitrary</u> when the broader <u>sense</u> of receptor is used (where there is no <u>implication</u> of <u>transduction</u> of <u>signal</u>). In these <u>cases</u> it is <u>probably</u> a good <u>rule</u> to consider the ligand to be the smaller of the two <u>thus</u> in a <u>lectin sugar interaction</u>, the sugar would be the ligand (<u>even</u> though it is attached to a much larger molecule, recognition is of the saccharide).

#### Exerting Forces on Single Molecules – Single Particle (Cell / Vesicle / Bead)



#### ... or any combination

• . . .



- •Force between single ligand receptor pairs
- •Cell adhesion on specific surfaces
- •Folding/Unfolding transition of polypeptides
- •Elasticity and structure of chromosomes by aspiration



Meaningful data acquisition can be long and tedious since good statistics are required

# Adhesion Plays an Important Role in Biology

### Adhesion in Pathogenesis



# Importance of Adhesion Studies

- Basic studies of adhesion provide understanding of mechanisms governing binding
- Can lead to therapeutic advances by adhesion inhibition or promotion
- Most studies are equilibrium studies of binding to surfaces, leaving out dynamics that may be crucial
  - surface flow measurements are difficult because of velocity uncertainty at surfaces

# **Types of Adhesion**

- Non-biospecific
  - hydrophobic or electrostatic
- Biospecific and exclusively adhesive
- Bioespecific and functional
  - integrins to RGD or fibronectin
  - selectins binding to cadherins

# Simplest Specific Adhesion View

- A pathogen (lock) selectively binds to a suitable receptor (correct key) on the surface of the cell, while not accepting incorrect binding site (wrong key)
- polyvalent binding is more specific



# Hemagglutination Assay for Virus Inhibitor





# Adhesion Can be Dynamic

# White blood cells adhesion to blood vessel walls at onset of inflammation



# Spontaneous Unbinding of Single Bonds Occurs

- polyvalent binding decreases unbinding probability by making deeper well
- if a single bond unbinds, other remaining bonds can hold pathogen to cell allowing rebind



# Dynamical Control of Collisions Desirable

- AFM possible
- We choose optical tweezers



# Optical Tweezers are an Excellent Tool for Research on Adhesion

- Tweezers based manipulation can tailor the position, orientation, velocity, duration and force of a contact between surfaces
- Tweezers do not require that anything be attached to the cell being manipulated
- Tweezers can measure the force required to separate surfaces under controlled mechanical conditions
- Adhesion not previously considered for tweezer measurements because binding forces >> tweezer forces
  - breakthrough idea: tweezers do not have to separate bound particles, but just to distinguish bound from unbound

# Optical Tweezers Potential = $-\alpha$ Intensity = $-\beta$ | Electric Field |<sup>2</sup>

1. Establish an Electric Field Maximum by focusing a light beam



2. Particles will be attracted to the light beam focus where the Electric Field is a Maximum



# **First Optical Tweezer**



# Measurements of Adhesion Between a blood cell or bacteria and an artificial surface

- Used to measure pathogen bonding
- Dynamics of binding can be probed
- Used to measure inhibitors for pathogen binding

# First Optical Tweezer Based Adhesion Experiments

(CFLD by Phillips and Whitesides Groups)

- Multi-beam optical tweezer controls position of two or more particles
- Adhesion between particles is measured by determining the tweezer force required to pull the objects apart.
- First experiment was on adhesion of influenza virus to red blood cell
- Studied effects of inhibitors
  - important for drug development where effective inhibition at lowest possible concentration is desired
- Tweezers too weak to pull apart spheres bound to cells by virus, but easily pull apart unbound cells and spheres

# Dual Optical Tweezers (DOT) For Biochem



#### OPTCOL to Measure Inhibition of Adhesion by Soluble Carbohydrates





#### OPTCOL to Measure Inhibition of Adhesion by Soluble Carbohydrates





# **Dual Tweezer Controlled Collision**

#### **Inhibited Adhesion**

coated sphere



erythrocyte



#### erythrocyte

Ten trials. Adhesion in <3 is counted as binding, no adhesion > 10 no binding. 3-10 rare.

# Results of First Tweezers Adhesion Experiment

- Dot used to measure effectiveness of viral anti-binding agents
  - Passivation of bead surface vastly increases specificity of test
  - 50% adhesion point matches standard assay in range where standard is valid (>10<sup>-9</sup> molar)
  - Extends dynamic range 2 orders of magnitude beyond standard test (10<sup>-11</sup> to 10<sup>-3</sup> molar)
  - Agents indistinguishable in standard assay shown to have orders of magnitude different effectiveness



2

# Measurements of Cell Binding to Pathogen Surrogates

- Use a molecule with known structure to probe binding
- Provide detailed comparison of theory and experiment

# Second Model System WGA and Erythrocyte

#### Wheat Germ Agglutinin

 well characterized dimeric lectin that binds to GlcNAc and NeuAc with four binding sites



Figure 11. Ribbon representation of wheat germ agglutinin (PDB entry WGC). The sulfur atoms of the disulfide bridges are shown as small balls.



Figure 12. Schematic representation of the wheat germ agglutinin dimers. Domains are shown as large shadowed circles and labeled A1, B1, C1, D1, etc. The position of the molecular 2-fold axis is indicated by an arrow. Broker arrows represent the two types of pseudo-2-fold axes generated in the dimer interface between domains or different dimers. "S" refers to the aromatic carbohydrate binding pocket. (Reprinted by permission from ref 51

# Inhibition of WGA binding by Soluble Sugar



# WGA binding results

- Uncoated spheres stick to each other and to erythrocytes
  - BSA blocks uncoated adhesion
  - Not biospecific
    - » electrostatic or hydrophobic
- EG coated spheres do not stick to each other or to cells
- In the absence of inhibitor, WGA coated spheres always stick to cells and could not be removed with tweezers
- GlcNac and sialic acid in solution inhibit the binding
  - inhibition concentrations for 50 % sticking are similar to concentrations required to inhibit hemagluttination
- Other soluble carbohydrates do not block binding
  - glucose had no effect
- No Spontaneous unbonding observed

# Bacteria Binding to an Artificial Surface

- Can precisely control surface chemistry using self-assembled monolayers
- Time dependent responses dominated by bacterial changes rather than surface changes
  - over long times remodeling of surface is still an issue
- Model system: E. coli specifically binding to mannose

– clinical sample



Adhesion to cells is frequently really adhesion to sugars on the outside of the cells. Sugars can encode far more info than amino acids, so cells are often identified by the sugars on their surfaces.

# Microbial Infection begins with adhesion to a sugar

Influenza virus - "Flu" HIV - AIDS Helicobacter pylori - Ulcers Escherichia coli - Meningitis Pseudomonas aeruginosa - Pneumonia Trypanosomes - African sleeping sickness Plasmodium falciparum - Malaria

# Chronic inflammatory disease begins with adhesion to a

### sugar

Diabetes Multiple sclerosis Rheumatoid arthritis Inflammatory bowel disease Psoriasis Transplant rejection Asthma

# Inhibition of Adhesion with minimal dose can be therapeutically important

- Adhesion minimization can prevent infection and inflammation
- Smallest possible doses reduce side effects
- Lectins are proteins that bind mono- and oligosaccharides reversibly and with high specificity
  - not catalysts
  - not products of immune response
- Lectins frequently cause specific adhesion
- Search for inhibitors that bind to the lectins preventing pathogens from binding to cells
- Need quantitative method of evaluating inhibitors
  - measure equilibrium binding in solution
  - no dynamical information

# Tweezer Based Measurements of E. Coli Binding

(Michael Liang and Stephen Smith)

- Goal: Measuring the specific binding of E.coli to mannose
- Strategy: Control the density of mannose on a surface using self-assembled monolayers (SAMs)



# **Apparatus Detail**



# Initial Steps in Adhesion Measurement



# **Steps in Adhesion** Measurement

Place Low Power Trap Below Adhering Bacteria



**Tweezers Force** 

Increase Trap Strength Until Bacteria Unbinds because Tweezer Force = Bond Rupture Force Sugar Coated



No Net Average Tweezers Force

# Effect of Multiple Binding Sites on Binding Force





Binding force = N single bond for N on single site Binding force < N single bond for multiple site




### Tweezer Based Measurements of E. Coli Binding

- First observation of unbinding with optical tweezers
- Demonstrated quantized specific binding of E. Coli to surfaces
- Measured effects of inhibitors on binding force
  - some inhibitors INCREASED strength of individual bonds
- Found that each pili had at most two binding sites
  - consistent with literature of dimer at end of pili with only 1 dimer/pili available
- Found negligible probability of spontaneous release of doubly bound E.coli
- Observed spontaneous release of singly bound E. coli

can measure time between unbinding events

 Mannose in solutions will cause bound E.coli to unbind even without tweezers

### Single Molecule Binding of Ligand/Receptor Pairs isolated on artificial surfaces

- Determine the energy landscape of the binding potential
- Measure the dynamics of binding and unbinding
- Measure the effects of inhibitors
- Measure steric effects
- Measure effects due to the presence of macroscopic surfaces

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### Quantitative Information K<sub>on</sub> K<sub>off</sub>

#### Law of Mass action

According to the law of mass action, the reversible reaction A + B ? AB can be described as two reactions that occur at characteristic rates that depend on the energy landscape of the bond, reactant concentration, and temperature only

$$d[AB]/dt = k_{on}[A][B]$$

$$d[A]/dt = d[B]/dt = k_{off} [AB]$$

where k<sub>on</sub> and k<sub>off</sub> are constant functions of temperature



### **Effect of Applied Force**

#### According to the Bell model,

W = -F<sub>magnetic</sub> d [] G(F) = [] G(0) - F x

where x is the reaction coordinate, which, in a simple model, is roughly the distance that A and B must be separated to break the bond.



### Measuring K<sub>off</sub> (0) and x using Applied Force

- The off rate, k<sub>off</sub> (F) is measured by examining the number of unbroken bonds as a function of time:
- N(t) = N(0) exp(-t k<sub>off</sub> (F))
  - By measuring the  $k_{off}$  (F) rate at various forces, it is possible to extrapolate the  $k_{off}$  rate at zero force ( $k_{off}$  (0) or  $k_{off}$  in solution) as well as to calculate the unbinding reaction distance (x) and the transition state barrier (DG). The graph of F vs  $k_{off}$  (F) can be fitted to the decay equation:
- $k_{off}(F) = k_{off}(0) \exp(F x/k_b T)$
- If more than one type of bond is present, or there is more than one minima in the potential then multiple decay times will be present, each of which can be treated as described above.

### **AFM Results** (force depends on time)

 $\begin{aligned} k_{off} &= v_{off} \exp(-\Box G_{off} / k_b T) \\ k_{off} (F) &= k_{off} (0) \exp(F x / k_b T) \end{aligned}$ 

#### Measure rupture force and distance





### **AFM Results** distribution of rupture forces as a function of loading rate

198 pN/sec

2300 pN/sec



### **AFM Results** evidence for an intermediate state



Reaction coordinate

FIGURE 4: Conceptual energy landscapes of the (A) streptavidinbiotin interaction and the (B) W120F-biotin interaction.



FIGURE 3: Loading rate dependence of the rupture force in the unbinding of the streptavidin—biotin (○), avidin—biotin (□), and W120F—biotin (●). Both regimes in force spectra were fitted to the Bell model. Standard errors of all data points were less than 5% of the mean value. Representative error bars were placed on selected data points.

#### Ligand-Receptor Binding using Magnetic Tweezers at constant force



Single molecule measurements of ligand- receptor interactions.
Compare with other techniques (relate single molecule results to bulk properties).

•Probe energy landscape.

•Surface confined molecules.

•Good statistics possible with parallel measurements.

### **Magnetic Crystals** of Paramagnetic Beads Energy= |m||B|



### Beads Attached to a surface are pulled away, exerting a force on the molecule attaching the bead to the surface



### Magnetic Tweezer Based Adhesion Measurements

#### •Experiment:

Allow the beads to contact the surface for 30".

Apply a low force ~ 5-10 pN

Apply force required to unbind ligand- receptor pairs.



### Negative control



#### Positive control after low force





# **Non-Specific Binding**

- A big issue in early experiments
- A big problem for bead based assays
- Can combat with appropriate surface coatings
- Inert to protein binding not the same as inert to bead binding

### **Different "Inert" Surfaces**

SURFACE	BEADS	SOLUTION	LOW FORCE	HIGH FORCE	CONTRO L
BSA on PVC	Streptavidin	Buffer PBS	40-60 % beads remaining		(-)
BSA on PVC or PVC	Streptavidin	Buffer PBS / Tween 20, 0.05%	< 5 % beads remaining		(-)
Biotinylated-BSA on PVC	Streptavidin	Buffer PBS / Tween 20, 0.05 %	90 % beads	40 – 80 % remain depending on force	(+)
Biotinylated-BSA on PVC	Streptavidin and incubated with biotin 10 mM – 1 μM	Buffer PBS / Tween 20, 0.05 %	< 5 % beads remaining		(-)
Biotinylated-BSA on PVC and incubated with streptavidin	Streptavidin	Buffer PBS / Tween 20, 0.05 %	< 5 % beads remaining		(-)
BSA : Biotinylated-BSA: 500 :1, on PVC	Streptavidin	Buffer PBS / Tween 0.05 %	< 15 % beads remaining		(-)
Streptavidin on PVC or PVC	Biotin PEO- or biotinPEO + streptavidin	Buffer PBS / Tween 0.05% or + biotin 10 mM	< 5 % beads remaining		(-)

### Results



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-6

-7

-8

-9

-10

0

10 20

30 40 50

Force (pN)

60 70 80

### Comparison with Other Techniques

Biomolecular Recognition at Solid-Liquid Interfaces			J. Am. Chem. Soc., Vol. 121, No. 27, 1999 6477						
Table 1. Apparent Kinetic Constants of Dissociation, $k_1^{supp}$ , for Streptavidin Mutants Bound to Biotin-Terminated SAMs <sup>a</sup>									
$k_1^{\text{app}}(s)^{-1}$									
$k_{\text{off}}(s^{-1})$		SAMs formed from 2 and 3							
solution <sup>24,27</sup>	SAMs formed from $\mathbf{l}$	$\chi_2 = 0.15$	$\chi_2 = 0.007$						
$(3.3 \pm 0.1) \times 10^{-6}$ $(5.7 \pm 0.1) \times 10^{-4}$ b	$(4.16 \pm 0.17) \times 10^{-4}$ $(5.40 \pm 0.08) \times 10^{-2}$	$\begin{array}{c} (1.37\pm0.08)\times10^{-4} \\ (2.05\pm0.04)\times10^{-3} \\ 0.336\pm0.019 \end{array}$	$(3.31 \pm 0.16) \times 10^{-4}$ $(7.25 \pm 0.40) \times 10^{-2}$ $2.44 \pm 0.24$						
ti e	tic Constants of Dissociation $ \frac{k_{\text{off}}(s^{-1})}{\text{solution}^{24,27}} $ (3.3 ± 0.1) × 10 <sup>-6</sup> (5.7 ± 0.1) × 10 <sup>-4</sup> b		$ \begin{array}{c} \mbox{ion at Solid-Liquid Interfaces} & J. Am. Chem. Soc., Vol. \\ \mbox{tic Constants of Dissociation, $k_1^{\rm UP}$, for Streptavidin Mutants Bound to Biotin-Terminated} \\ \hline \\ $						

<sup>a</sup> The rate constants of dissociation of streptavidin-biotin in solution are smaller than on the surface and were obtained from refs 24 and 27. <sup>b</sup> Dissociation of biotin from W120A is too fast (<60 s) for an accurate determination of k<sub>off</sub>.

10222 Biochemistry, Vol. 39, No. 33, 2000				AFM: Moy					
Table 1: Bell Model Parameters from $f^*$ vs log $(r_f)$ Relation									
ligand—receptor pair	loading rate range (pN/s)	$x_{\beta}$ (nm)	$k^{\circ} (s^{-1})$		$\Delta\Delta E^a$ ( $k_{\rm B}T$ )				
streptavidin-biotin	100 - 1000 1000 - 5000	0.49 0.05	1.67 > 2.09	< 10 <sup>-5</sup>					
avidin-biotin	100-1000 1000-5000	0.53	6.45 > 0.08	< 10 <sup>-6</sup>	0.95 3.26				
W120F-biotin	100-1000 1000-5000	0.31 0.11	6.70 > 1.05	< 10 <sup>-3</sup>	-5.99 0.69				
$^{a}\Delta\Delta E$ is relative to streptavidin-biotin binding energy.									

These results: avidin  $k_{off} = 9.10^{-5}$ , x= 0.36 nm; streptavidin  $k_{off} = 1.4.10^{-4}$ , x= 0.38 nm.

BFP (Evans, 1999): avidin- biotin 38 to 85 pN regime: x=0.3 nm.

Results for CA on SAMs also showed higher k<sub>off</sub> values for surface attached ligand-receptor complexes (Whitesides JACS 117, 12009).

# **Guaranteed Single Molecule**



Use a long spacer with only one single binding site

Long linkers on both sides allow better comparison with solution values

### Summary of Ligand-Receptor Binding using Magnetic Tweezers

#### Accomplishments

✓ Surface: designed to reduce non-specific interactions significantly (<1%).

✓ Interaction studied: biotin- avidin (streptavidin).

✓ Calculation of parameters at constant force: reaction off rate, unbinding reaction distance.

#### **Future Work**

 Include tethers of different lengths to approach solution conditions.

•Other surfaces: gold.

• Study other ligand- receptor couples such as sulfonamidescarbonic anhidrase or ligand (receptors) on cell surfaces.

### Unzipping Lambda phage dsDNA using magnetic tweezers



### What is the physics?

Unzipping occurs when the free energy of the single stranded is lower than the free energy of the double stranded

Contribution due to entropy difference favors single stranded

Contribution do to enthalpy difference favors double stranded

Applied force aids separation

 $\Box \mathbf{G} = \mathbf{g}_b - \mathbf{2} \mathbf{g}_u (\mathbf{F})$ 

# Observations of the Sequence Dependence of dsDNA unzipping



# **Comparison with Theory**

#### Averaged Potential Theory did not work well Monte-Carlo simulation is a good match



#### Accomplishments

 ✓ Pauses in force-induced unzipping of double stranded were successfully predicted by Monte Carlo, but not by coarse graining

#### **Future Work**

Investigate interactions
 beyond nearest neighbors

•Explore the kinetics of unzipping through simulations

### Temperature Dependence of Unzipping

#### □ G= □ H – T □ S

unzipping should get easier at higher temperatures

DH and DS assumed independent of temperature

# Present results as a phase diagram in the force temperature plane

at zero force the transition temperature should be the melting temperature

# Allows comparison of theory with experiment

temperature dependence important for DNA chips, PCR, and projection of melting data to in vivo conditions

# Phase Diagram of DNA Unzipping



•Surprises at high and low temperature, dependence on thermal history

### **New Biophysics?**

- In the high temp low force regime effects of bubbles and hairpins have been ignored
  - detailed simulation by R. Bonshudt suggests hairpins do not play a role
  - Libchaber data supports bubbles, but this is not yet widely held
- In low temp regime a structural change appears to take place
  - supported by CD data
  - supported by force vs extension data
  - supported by history dependence
  - consistent with previous short sequence data on B' to B transition

### More new biophysics

- Unzipping force is buffer insensitive from 20-40 C
  - stretching curves change drastically
  - melting temperature changes of 30 C
- Biological robustness makes sense, detailed mechanism still not understood

# Summary Phase Diagram of DNA Unzipping

#### Accomplishments

✓ Between 24-35°C good projections from bulk thermodynamic using nearest neighbor, insensitive to buffer
 ✓ Above 35°C unzipping force depends on buffer and bubbles in the dsDNA may be important
 ✓ Below 24°C dsDNA conformational changes may play a role
 ✓ Demonstrate that dsDNA parameters depend on previous history of the sample

#### **Future Work**

•Evaluate effect of sequence dependent conformation (e.g. B' or A tract)

- Improve predictive models for PCR primers
- •Develop rapid lower temperature PCR
- Understand history dependence of dsDNA parameters

### Magnetic Field Induced Lattices

**Goal:** Assemble small particles into an ordered hexagonal lattice

**Strategy:** Use a magnetic field gradient to attract paramagnetic particles to a center, and dipole/dipole interaction to keep them slightly separated





The beads form an ordered crystal with a spacing increasing as |B|/(grad|B|)

### It is often Desirable to Join Long DNA Strands



Random walk is not sufficient to bring ends together increasing concentration makes a gel the rate of end attachment DECREASES

# Lattice Based Assembly

**Goal:** Attach two separate double strands of DNA to form one longer double strand, and detect successful assembly

**Strategy:** Attach the DNA to magnetic beads, form beads in crystal, detect DNA attachment as crystal defects





Random walk is not sufficient to bring ends together increasing concentration makes a gel the rate of end attachment DECREASES



DNA concentration is low, but probability of joining ends is high.

### Examples



### Join DNA Ends by Bringing Beads Together

DNA concentration is low, but probability of joining ends is high.

### Magnetic tweezers

#### Self-Assembled Crystal



In the presence of a vertically oriented magnetic field, micron-sized superparamagnetic beads form a 2-Dimensional crystal at the surface of a droplet of water. The spacing between beads can be adjusted by varying the magnetic field strength. We image the crystal using video





#### Binding Properties of Large Molecules

In order to study the properties of long polymers (such as DNA), we chemically attach the molecules to the beads. Then, by controlling the crystal spacing, we control the time and distance at which the polymers interact with each other and with other beads..





#### Crystal Defects





#### **Bound Beads**

Beads bound to each other by a polymer show up as defects in the crystal lattice. Using digital image processing, we can then measure the dynamics of the bound beads under various conditions (crystal spacing, temperature, fluid flow), and extract useful information about the polymer that connects them.





### Results of Lattice Based Measurements

- Most probable binding difference
   radius of gyration
- Highly peaked distribution

### Adhesion For Microfabrication

- Biospecific interactions can be used to build structures from combinations of cells and nonliving material
- Useful for creating and studying bio-materials interfaces
- Methods for creating microstructures that can be disassembled at will
#### Macrofabrication: Strategies for Microfab



#### Essential Research Areas for Developing Functional Devices with Biological Components

Materials for Controlling the Interactions of Cells and Proteins with Synthetic Surfaces

Strategies to Control the Environment of Cells in 2D and 3D Arrays and Assemblies

Strategies to Detect and Transduce Biological Signals and Responses

#### **Multiple-Beam Optical Tweezer to Orient Ery**





### Erythrocytes and Lymphocytes in One Assembly



# Light-Driven Microfabrication using WGA coated Spheres



- A) Attach sphere to erythrocyte
- B) Attach erythrocyte and sphere to growing assembly that rests on the surface of the cover slip
- C) Iterate to generate 2D and 3D arrays





#### **Fabrication of 2D Arrays of Erythrocytes**



#### 3D Fabrication: A Tetrahedral Array of Erythrocytes



#### **3D Fabrication: Perpendicular Planes**



#### Ligand-Receptor Binding using Magnetic Tweezers Whitesides

Groups



#### **Magnetic Cell Organization**

Ingber and Prentiss Groups



#### Phase Diagram of DNA Unzipping

Ingber, Nelson, Prentiss Groups



#### Large force over small Area Ingber, Prentiss, Whitesides, Xia Groups



#### Ligand-Receptor Binding using Magnetic Tweezers Whitesides

Groups



#### **Magnetic Cell Organization**

Ingber and Prentiss Groups



#### Sequence Dependence of dsDNA unzipping Ingber, Nelson, Prentiss Groups



#### Large force over small Area Ingber, Prentiss, Whitesides, Xia Groups



### **Gradients and Dipoles**

#### **Opposite and Equal Pull -> No Motion**



**Opposite and Unequal Pull -> Motion** 



## **Gradients and Dipoles II**

#### Negative q in E



#### Positive q in E



Dipole (Attached Negative and Positive q)



#### **Opposite and Equal Pull -> No Motion**



### **Gradients and Dipoles III**



**Opposite and Equal Pull -> No Motion** 



#### **Non-uniform Field**



#### Opposite and Unequal Pull -> Motion toward Increasing Field









### Potential= - [] | Electric Field |<sup>2</sup>

same origin as optical tweezers potential

1. Establish an Electric Field Maximum at the comb by charging



2. Particles will be attracted to the comb where the electric field is a maximum and potential is a minimum



3. Particles at the potential minimum can be manipulated by moving the comb, even if they are too small to hold in a hand



# Simple Tweezers



### Microscope

Cost ~ \$5k

### Light Assisted Self-Assembly (Patrycja Paruch)

- Goal: Deposited carbon nanotubes across a gap between two conductors for room temperature Single Electron Transistor
- Strategy: Use light force enhancement at conductor edge to attract nanotubes to gates
- Achievements: Deposited nanotubes by centering tweezer on the edge a conductor



stc: 95

### **Scaling Laws**

- Ball bearings self assemble under gravity < 10 micron particles do not
- Depends on assumptions
- For constant magnetization
  - Surface B field [] r
  - Field gradient is constant at the surface
  - magnetic moment of bead a r<sup>3</sup>
  - External B field dipole a r3
  - Dipole/Dipole

### **Magnetic Crystals** of Paramagnetic Beads Energy= |m||B|



### Magnetic Crystals of Paramagnetic Beads Energy= |m||B|



Surface Tension Keeps Beads in the water, so the beads accumulate at the surface, at the points where |B| is a maximum . If B is perpendicular to the water surface, the dipole moments are parallel, so neighboring beads repel.

#### BSA modified with biotin on PVC (streptavidin (avidin) coated beads).





#### Beads modified with biotin (streptavidin (avidin) on PVC).



•EZ-Link® Biotin-LC-PEO-Am Spacer arm: 22.9Å

#### Dynabeads® M-450 Tosylactivated





- Hydrophobic bead.
- Surface tosyl groups.
- Bead diameter 4.5 µm.

 Direct covalent binding to primary amino- or sulfhydryl groups in proteins and peptides.

- No further surface activation required.
- Binding over night at neutral to high pH and high temperature.

### Conjugation of WGA to Polystyrene Microspheres









Energy Landscape of Streptavidin-Biotin Complexes



FIGURE 2: (A) Force vs displacement curves of the interaction between a streptavidin-functionalized tip and a biotinylated agarose bead. The measurement recorded the force on the AFM cantilever on approach and retraction of the cantilever from the agarose bead. f<sup>\*</sup> is the rupture force.  $k_S$  is the slope of the force vs displacement curve. The measurement was carried out with a cantilever spring constant of 30 mN/m and a scan speed,  $v_c$ . of 600 mm/s. Each cantilever was individually calibrated by thermal fluctuation analysis to determine its spring constant (26). Histograms of the adhesion force between a streptavidin tip and  $v_c = 86 \text{ mm/s}$ ) and (C) 2300 pN/s ( $k_S = 2.3 \text{ mN/m}$  and  $v_c = 1 \ \mu \text{m/s}$ ). Both histograms were fitted to a Gaussian function (2). The centers of the force distribution (B) and (C) are  $126 \pm 2.3 \ (\text{SEM})$  pN (N = 256) and  $207 \pm 5.8$ (SEM) pN (N = 100), respectively. All AFM force measurements were carried out in PBS and at  $25 \pm 1$  °C.

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FIGURE 3: Loading rate dependence of the rupture force in the unbinding of the streptavidin—biotin (O), avidin—biotin ( $\Box$ ), and W120F—biotin ( $\bullet$ ). Both regimes in force spectra were fitted to the Bell model. Standard errors of all data points were less than 5% of the mean value. Representative error bars were placed on selected data points.